# NONPROVISIONAL PATENT APPLICATION TRANSMITTAL RULE §1.53(b) IN THE UNITED STATES PATENT AND TRADEMARK OFFICE



TO ARENT FOX KINTNER PLOTKIN & KAHN, PLLC

1050 Connecticut Avenue, N.W.,

Suite 600

Washington, DC 20036-5339 Telephone: (202) 857-6000 Facsimile: (202) 638-4810

Commissioner for Patents Washington, D.C. 20231

Sir:

 $\boxtimes$ 

Transmitted herewith for filing under 37 C.F.R. §1.53(b) is a nonprovisional patent application:

PROCESS FOR ADVANCED TREATMENT FOR DESULFURIZATION OF For (Title):

EXHAUST GAS WITH SEAWATER

Kouji SHIRAISHI; Takayoshi HARIMOTO; Toshiaki MATSUOKA: By (Inventors):

Naoki FUJIHATA; Akihiko HONGYOU; Katsuo OIKAWA; Kazuo TAKEDA

Date:

13 pages of Specification is attached.

Drawings (4 sheets) is attached.

A Declaration and Power of Attorney is attached.

An assignment of the invention to \_\_\_FUJIKASUI ENGINEERING CO., LTD. \_\_ is attached,

along with Form PTO-1595 and a check for \$40.00.

The Applicant is a Small Entity.

A certified copy of Japanese application(s) No.(s) 11-312466; dated November 2, 1999

The filing fee is calculated below and includes claim status after entry of any Preliminary Amendment noted above:

FOR:	NO. FILED	NO. EXTRA
BASIC FEE	120	
TOTAL CLAIMS	2 - 20	= 0
INDEP CLAIMS	2 - 3	= 0
☐ MULTIPLE DEPENDENT CLAIMS		

SMALL ENTITY			LARGE I	ENIIIY	
RATE	FE	E	<u>OR</u>	RATE	FEE
Washie	\$ 35	5	<u>OR</u>		\$ 710
x 9 =	\$	0	<u>OR</u>	x 18	
x 40 =	\$	0	<u>OR</u>	x 80	
+135 =	\$	0	<u>OR</u>	+270	
TOTAL	\$ 35	5	OR	TOTAL	

M A Check No. 303840 in the amount of \$395.00 is attached. Please charge any fee deficiency or credit any overpayment to Deposit Account No. 01-2300.

Respectfully submitted,

George E. Oram, Jr. Registration No. 27,931

GEO/hk

10

15

20

25

30

35

# PROCESS FOR ADVANCED TREATMENT FOR DESULFURIZATION OF EXHAUST GAS WITH SEAWATER

#### BACKGROUND OF THE INVENTION

# 1. Field of the Invention

The present invention relates to a method of wet removal of an acidic component in an exhaust gas with seawater, in particular an acidic component exhausted from a boiler or various furnaces, and an acidic component from an exhaust gas containing sulfite gas.

# 2. Description of the Related Art

At present, for example, when heavy oil, coal, etc. are burned in combustion equipments such as a boiler or various types of furnaces and since the fuels contain a sulfur component, the sulfur component is combined with the oxygen in the air at the time of combustion to form sulfur oxides which end up included in the combustion exhaust gas. Such sulfur oxides travel with the gas from the source of generation to regions as much as several thousand kilometers away to cause phenomena such as acid rain or acid fog, polluting the air, water, soil, etc. over a vast region and having detrimental effects on human health.

However, along with the increased international interest in global environmental issues, global scale measures have been sought. In Japan, along with advances made in exhaust gas desulfurization or denitrogenation technologies etc., efficient boiler combustion technology has been accomplished. Substantially 100% antipolution measures are now taken at the source of generation. For example, as a method of wet removal of the acidic component such as the sulfur oxides contained in the gas, a packed column, spray column, bubble cap column, perforated or grid plate column without weir and downcomer etc. have been used to bring the gas to be treated and an alkali treatment solution countercurrently

into contact. That is, it can be said that the desulfurization treatment technology for environmental protection has reached the almost perfected stage. For example, a desulfurization efficiency of 90 to 99% has been obtained and commercialized. However, since calcium hydroxide, calcium carbonate, sodium hydroxide, magnesium hydroxide, etc. are used as the alkali treatment solution, there occur problem that, in addition to higher cost, the waste solutions have to be treated, the solid substances have to be treated, etc. Thus, there are problems of complication of the process and high construction and running costs. Therefore, as a method of treatment of exhaust gas using seawater, the present inventors developed a method of wet treatment of exhaust gas by using a perforated or grid plate column without weir and downcomer in JP-A-11-290643 (method of treatment of acidic component in an exhaust gas with seawater). Nevertheless, this method was insufficient in the low Ug region.

As explained above, even a process for removing an acidic component from an exhaust gas, developments of new technologies for simplification of the process, making the equipment more compact, and sharply reducing the construction and running costs are sought not only in the industrialized countries, but in the developing countries as well. In particular, a reliable process is sought for cases where advanced treatment of all regions, including the low Ug region, is frequently required.

SUMMARY OF THE INVENTION

Accordingly, objects of the present invention are to develop new technologies capable of highly removing an acidic component such as a sulfate gas contained in an exhaust gas with a simplified process in a more compact equipment, and at a remarkably reducted the construction and running costs, in view of the international needs, at a wide region including a low Ug and low L/G region.

In accordance with the present invention, there is

25

5

10

15

20

30

35

COTACL" EZENGOSO

5

10

15

20

25

30

35

provided a method for removing an acidic component contained in an exhaust gas comprising by using a system comprising (a) a gas-liquid contact apparatus composed of an absorption column provided internally with at least one perforated plate at the top, bottom, or both top and bottom of the absorption column packed with at least one type of fillers, (b) an apparatus for introducing seawater to the absorption column, (c) an apparatus for oxidizing the seawater after gas-liquid contact, and (d) an apparatus for mixing a noncontact seawater with the seawater after subjected to the mixing and oxidation, whereby the exhaust gas containing an acidic component is brought into gas-liquid contact with the seawater.

In accordance with the present invention, there is also provided a wet method for removing an acidic component contained in an exhaust gas, in a gas-liquid contact apparatus including an absorption column having a column diameter of at least 500 mm and provided with at least one perforated plate having an free-space ratio Fc of 0.25 to 0.5 and provided with at least one type of packing material with a packing height of 0.5 m to 4 m, comprising

supplying seawater in such an amount that a ratio L/G of the flow rate L  $(kg/m^2\cdot hr)$  of the seawater to the flow rate G  $(kg/m^2\cdot hr)$  of the gas to be treated from the top of the column is at least 3.6 and a flow rate L of the seawater is 1 x 10<sup>4</sup> to 25 x 10<sup>4</sup> kg/m<sup>2</sup>·hr and introducing a treated gas in such an amount that a range of a superficial gas velocity Ug in the apparatus from the bottom of the gas-liquid contact apparatus is 0 to  $2\cdot Ugm$  (m/sec):

in the case of using a perforated or grid plate column without weir and downcomer composed of at least one perforated plate and the ratio  $\rho_{\rm c}/\rho_{\rm L}$  of the density  $\rho_{\rm c}$  (kg/m³) of the treated gas to the density  $\rho_{\rm L}$  (kg/m³) of seawater of 1030 is at least 0.838 x 10°3.

10

15

20

25

30

35

Ugm = 49.14 Fc<sup>0.7</sup>  $(\rho_{\rm g}/\rho_{\rm L} \times 10^{-3})^{-0.5} \cdot ({\rm L/G})^{-1/3} \cdot \sqrt{{\rm g \cdot L}}$ 

wherein L is a capillary constant  $\sqrt{2\sigma/\rho_{r} \cdot g}$ ,

g is a gravitational acceleration  $(m/sec^2)$ , and

 $\sigma$  is a surface tension of seawater (kg/sec<sup>2</sup>)

whereby the gas to be treated and seawater are countercurrently brought into gas-liquid contact.

BRIEF DESCRIPTION OF THE DRAWINGS

The present invention will be better understood from the description set forth below with reference to the accompanying drawings, wherein:

Fig. 1 is a schematic view showing the relationship between the flow rate L of the treatment liquid and the superficial gas velocity Ug in the operating region according to the method of the present invention, wherein the regions A and B show the operating regions according to Japanese Examined Patent Publication (Kokoku) No. 51-31036 and Japanese Examined Patent Publication (Kokoku) No. 60-18206, while the region C shows the operating region according to JP-A-11-290643. The operating region in the present invention is the region of Ug of more than 0, but not more than 2 x Ugm, surrounded by the lines of the flow rate of more than 1 x  $10^4$  (kg/m²·hr), L/G=3.6, and the flow rate 25 x  $10^4$  (kg/m²·hr), of the treatment liquid of Fig. 1.

Fig. 2 is a view of the treatment system centered around the apparatus for removal of the acidic component in the exhaust gas with seawater, that is, a gas-liquid contact apparatus, according to the present invention.

Fig. 3 is an example of an absorption apparatus consisting of a combination of a perforated or grid plate without weir and downcomer and a packed column in the case of treating boiler exhaust gas by the method according to the present invention.

Fig. 4 is another example of an absorption apparatus consisting of a combination of a perforated or grid plate without weir and downcomer and a packed column in the

10

15

20

25

30

35

case of treating boiler exhaust gas by the method according to the present invention.

DESCRIPTION OF THE PREFERRED EMBODIMENTS

According to the present invention, as shown in Fig. 2, since an absorption column composed of a combination of a perforated or grid plate without weir and downcomer and packing material is used, operation is possible from a small liquid/gas ratio to a large liquid/gas ratio, the alkali contained in the used cooling seawater or alkaline waste seawater used in power plants, factories, etc. can be utilized, and the desired acidic component can be removed, without using any chemicals at all, for the purpose of absorbing the gas and of adjusting pH of the absorbing liquid, in the entire process. Note that here the perforated or grid plate without weir and downcomer means a perforated or grid plate having a free-space ratio Fc of 0.25 to 0.5, preferably 0.3 to 0.4. In the present invention, by using a gas-liquid contact apparatus comprising a combination of at least one such perforated or grid plate and packing material with a packing height of at least 0.5 m, preferably 0.5 to 4 m, it is possible to use seawater or waste seawater to remove, in a wet manner, acidic components such as sulfite gas contained in an exhaust gas. Further, since the seawater after the gas-liquid contact makes the components absorbed into it harmless by the supply of air and oxidation by a seawater treatment system, there is no problem with pollution of the ocean at all.

The present invention is not limited in content to the following mechanism, but the chemical principle of the seawater desulfurization process according to the present invention is now explained, when the acidic component is sulfite gas (SO<sub>2</sub>).

That is, the sulfite gas  $(SO_2)$  contained in the exhaust gas is absorbed in the seawater, and then converted to bisulfite ions as shown in formula (1):

$$SO_2 (gas) + H_2 O = HSO_3^- + H^+$$
 (1)

10

15

20

25

30

35

The bisulfite ion thus formed is converted to a sulfate ion as shown in formula (2) by aeration exidation:

$$HSO_3^- + 1/2O_2 = SO_4^{2-} + H^+$$
 (2)

The hydrogen ion thus produced by the reaction formula (2) is then reacted with a carbonate ion or bicarbonate ion contained in seawater and is thus neutralized as shown in the reaction formulae (3) and (4):

$$CO_3^{2-}+2H^+ = CO_2$$
 (gas and liquid) +  $H_2O$  (3)

$$HCO_3^+ + H^+ = CO_2$$
 (gas and liquid) +  $H_2O$  (4)

According to the present invention, the sulfite gas finally becomes a sulfate ion which is then dissolved into the seawater. The sulfur content in the ocean is said to be as high as  $10^{15}$  tons. This corresponds to about 2300 mg/L of a sulfate ion. The concentration of the harmless sulfate ion contained in the finally treated seawater effluent from the present process only increases by about several mg/liter, and therefore the increase in the sulfur content in the ocean is extremely slight.

Some of the sulfur derived from the sulfur oxides produced due to the use of fossil fuels in modern day industry is naturally recycled by passing through the atmosphere and returning to the earth or the sea in the form of acid rain. As opposed to this, the method of removal of the acidic component in a gas in the present invention is an environmentally friendly technique. According to the process of the present invention, it is possible to recycle sulfur to the ocean in this way by a short cut and effectively prevent the damages such as acid rain, air pollution.

As explained above, the present invention successfully uses a gas-liquid contact apparatus equipped inside thereof with an absorption column containing at least one perforated plate and at least one type of packing materials with a packing height of at least 0.5 m, preferably 0.5 m to 4 m (e.g., Raschig rings, pole

10

15

20

25

30

35

rings, Terralets, interlock saddles, etc.) and introduces seawater from the top of the gas-liquid contact apparatus or introduces seawater from the top of the gas-liquid contact apparatus so as to cause countercurrent gas-liquid contact with the gas to be treated and to use the alkali in the seawater to effectively and advancely remove the sulfur oxides contained in the exhaust gas.

Note that the technology for treatment of an exhaust gas using a perforated plate or grid plate column without weir and downcomer (i.e. "Moretana" column) is disclosed in Japanese Examined Patent Publication (Kokoku) No. 51-31036 and Japanese Examined Patent Publication (Kokoku) No. 60-18208 (or U.S. Patent Nos. 3892837 and 3941572) but the present inventors found that the operating regions A and B (see Fig. 1) already shown there is not suited for the treatment of exhaust gas utilizing seawater. The above-mentioned JP-A-11-290643 proposed the need for the ratio L/G of the flow rate of gas G supplied to the column and the flow rate L of seawater to be at least 3.6, preferably 7 to 25, and for the superficial gas velocity Ug passing through the "Moretana" column and the flow rate L of the treatment liquid to be the relation in the region C of Fig. 1, that is, in the range from more than 3.43 L-0.0807 • Ugm (m/sec) to 8 (m/sec). However, the outside of this range, particularly when Ug is too low, there is the problem that the efficiency of gas-liquid contact drastically falls and it becomes impossible to remove the acidic component. However, according to the present invention, by utilizing an absorption apparatus composed of a combination of a "Moretana" column and a packing column, the desired gasliquid absorption is effectively performed even in the region D of Fig. 1 and the highly efficient desulfurization treatment becomes possible.

Seawater contains approximately 110 to 130 mg/L, in terms of CaCO<sub>3</sub>, of alkali. The present invention makes highly effective use of seawater - which can be easily

10

15

20

25

30

35

obtained in large quantities in the area near the sea. According to the present invention, for example, in the case of a power station using seawater of the nearby ocean as cooling water, it is possible to reuse the seawater after the cooling, originally to be returned to the ocean, to treat the exhaust gas from a boiler and remove the sulfur oxides contained in the exhaust gas therefrom at a high desulfurization rate. Furthermore, according to the present invention even in the case of a factory producing magnesium hydroxide from seawater or a pulp and paper factory using seawater, it is possible to reutilize the waste seawater, which originally has to be retreated before discharge to the ocean, and to use the above gas-liquid apparatus to treat both the acidic exhaust gas and waste seawater.

According to the present invention, the sulfurabsorbed liquid is discharged into the ocean after the acidic sulfur-absorbed liquid containing bisulfite ion, which becomes a source of COD (i.e., chemical oxygen demand), is oxidized by air in the aeration oxidation vessel and after the pH of the mixing seawater is recovered by decarbonization, and therefore, it is possible to discharge the seawater into the ocean and recover the quality of the seawater without using any chemicals.

#### EXAMPLES

The present invention will now be further explained in detail by, but is by no means limited to, the following Examples.

### Example 1

Figure 3 shows an example of a flow of the process, where boiler exhaust gas is treated by the method of the present invention.

Combustion exhaust gas containing about 1000 ppm of sulfur oxides exhausted from the boiler 1 is introduced into an electric dust collector 2 to remove the dust, then the exhaust gas is passed through a gas-liquid

10

15

20

25

30

35

contact apparatus 3 provided with one perforated plate having a free-space ratio Fc of 0.3 and a packed portion having 2 m of a packing material (i.e., a ring type, dimensions: 100  $\times$  78  $\times$  32 mmH) for treatment, then dispersed therefrom, as a gas effluent, into the atmosphere. Inside the gas-liquid contact apparatus 3, the exhaust gas is introduced from the bottom, while seawater pumped out from the sea is introduced from the top of the gas-liquid contact apparatus 3, whereby the seawater is countercurrently into contact with the exhaust gas in the apparatus 3 and absorb and remove the sulfur oxides contained in the exhaust gas. The acidic waste seawater containing the bisulfite ion discharged from the bottom of the apparatus, as a liquid effluent, is oxidized with air in the air oxidation vessel 4 to oxidize the sulfite ion and to decarbonize the gas effluent, whereby the pH of seawater is recovered and, then the seawater is discharged to the ocean. The removal rate was 90 to 99% with a liquid gas ratio L/G of 5 to 10, a Ug of 1 to 2 m/sec, and a concentration of sulfur oxides in the gas to be treated of 10 to 100 ppm.

#### Example 2

Figure 4 is another example of the method of treating boiler exhaust gas by the method of the present invention and shows a flowchart of an example in the case of treatment while maintaining a high desulfurization rate corresponding to the low duty operation of a boiler.

Combustion exhaust gas containing about 800 ppm of sulfur oxides exhausted from the boiler 1 is introduced into an electric dust collector 2 to remove the dust, then the exhaust gas is passed through a gas-liquid contact apparatus 3 provided with two perforated plates having a free-space ratio Fc of 0.3 and a packed portion having 1.5 m of a packing material (i.e., ring type, dimension: 100 x 78 x 32 mmH) for treatment, then discharged into the atmosphere. Inside the gas-liquid contact apparatus 3, the exhaust gas is introduced from

DOLLOTE EZECONÓ

5

10

15

20

the bottom thereof, while the seawater pumped out from the sea is introduced from the top thereof, whereby the seawater is brought into countercurrent contact with the exhaust gas in the apparatus 3 and the sulfur oxides in the exhaust gas are absorbed and removed. The acidic waste seawater containing the bisulfite ion discharged from the bottom of the apparatus as a liquid effluent is oxidized and decarbonized by air in the air oxidation vessel 4, and the pH of seawater is recovered, then the seawater is discharged, as a liquid effluent, to the ocean. The removal efficiencies were 99.4%, 98.7%, 97.9%, 96.9%, and 92.5% with a flow rate of seawater L (5 x 10<sup>4</sup> kg/m<sup>2</sup>·hr)<sub>2</sub>Ug of 0.25, 0.5, 1, 1.5, and 2 m/sec and a concentration of sulfur oxides in the gas to be treated of 5, 10, 17, 25, and 60 ppm.

As explained above, according to the present invention, it is possible to advancely and effectively treat the sulfur oxides contained in an exhaust gas with seawater, despite the region being unable to be treated by "Moretana" column in the past, that is, a low Ug, and possible to protect the environment simply and with a compact equipment and low cost.

10

15

20

25

30

35

#### CLAIMS

- 1. A method for removing an acidic component contained in an exhaust gas comprising by using a system comprising (a) a gas-liquid contact apparatus composed of an absorption column provided internally with at least one perforated plate at the top, bottom, or both top and bottom of the absorption column packed with at least one type of fillers, (b) an apparatus for introducing seawater to the absorption column, (c) an apparatus for oxidizing the seawater after gas-liquid contact, and (d) an apparatus for mixing a noncontact seawater with the seawater after subjected to the mixing and oxidation, whereby the exhaust gas containing an acidic component is brought into gas-liquid contact with the seawater.
- 2. A wet method for removing an acidic component contained in an exhaust gas, in a gas-liquid contact apparatus including an absorption column having a column diameter of at least 500 mm and provided with at least one perforated plate having an free-space ratio Fc of 0.25 to 0.5 and provided with at least one type of packing material with a packing height of 0.5 m to 4 m, comprising

supplying seawater in such an amount that a ratio L/G of the flow rate L  $(kg/m^2\cdot hr)$  of the seawater to the flow rate G  $(kg/m^2\cdot hr)$  of the gas to be treated from the top of the column is at least 3.6 and a flow rate L of the seawater is 1 x 10 $^4$  to 25 x 10 $^4$  kg/m $^2\cdot hr$  and introducing a treated gas in such an amount that a range of a superficial gas velocity Ug in the apparatus from the bottom of the gas-liquid contact apparatus is 0 to  $2\cdot Ugm$  (m/sec):

in the case of using a perforated or grid plate column without weir and downcomer composed of at least one perforated plate and the ratio  $\rho_{\rm g}/\rho_{\rm L}$  of the density  $\rho_{\rm g}$  (kg/m³) of the treated gas to the density  $\rho_{\rm L}$  (kg/m³) of seawater of 1030 is at least 0.838 x 10<sup>-3</sup>.

Ugm = 49.14 Fc<sup>0.7</sup>  $(\rho_{\rm e}/\rho_{\rm L} \ x \ 10^{-3})^{-0.5} \cdot (L/G)^{-1/3} \cdot \sqrt{g \cdot L}$  wherein L is a capillary constant  $\sqrt{2\sigma/\rho_{\rm L} \cdot g}$ , g is a gravitational acceleration (m/sec<sup>2</sup>), and  $\sigma$  is a surface tension of seawater (kg/sec<sup>2</sup>) whereby the gas to be treated and seawater are

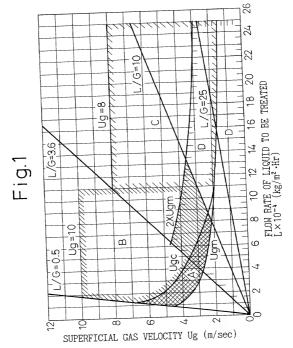
whereby the gas to be treated and seawater are countercurrently brought into gas-liquid contact.

# PROCESS FOR ADVANCED TREATMENT FOR DESULFURIZATION OF EXHAUST GAS WITH SEAWATER

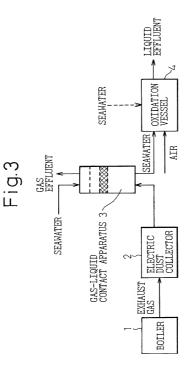
5

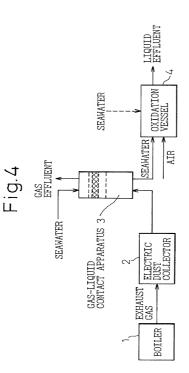
### ABSTRACT OF THE DISCLOSURE

A method for removing an acidic component contained 10 in an exhaust gas comprising by using a system comprising (a) a gas-liquid contact apparatus composed of an absorption column provided internally with at least one perforated plate at the top, bottom, or both top and bottom of the absorption column packed with at least one 15 type of fillers, (b) an apparatus for introducing seawater to the absorption column, (c) an apparatus for oxidizing the seawater after gas-liquid contact, and (d) an apparatus for mixing a noncontact seawater with the seawater after subjected to the mixing and oxidation, 20 whereby the exhaust gas containing an acidic component is brought into gas-liquid contact with the seawater.



➤ LIQUID ➤ EFFLUENT CHIMNEY SEAWATER TREATMENT SYSTEM ELIMINATOR BLOWER - COLUMN Fig.2 EXHAUST GAS 🗗 – SEAWATER -





# Declaration and Power of Attorney For Patent Application

# 特許出願宣言書及び委任状

La.1. ETK-41078-13

日本語宣言書		
下れの氏名の発明者として、私は以下の通り宣言します。	As a below narrd inventor, I hereby declar 'hat:	
私の住所、私書客、国籍は下記の私の氏名の後に記載され た通りです。	My residence, post office address and citizenship are as stated next to my name.	
下記の名称の発明に関して頂水道圏に記載され、特許出類 上でいる発明内容について、起が最初から唯一の発明者(下 間の氏名が一つの場合)もしくは最初かつ共同発明者である。 (下記の名称が複数の場合)信じています。	I believe I am the original. If ret and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural names are listed below) of the subject matter which is claimed and for which a patent is sought on the invention entitled	
-11 -12 -13 -14	PROCESS FOR ADVANCED TREATMENT FOR DESULFURIZATION OF EXHAUST GAS WITH SEAWATER	
・ ・ ・ ・ ・ ・ ・ ・ ・ ・ ・ ・ ・ ・ ・ ・ ・ ・ ・	the specification of which is attached hereto unless the following box is checked:  was filed on as United States Application Number or PCT International Application Number and was amended on (if applicable).	
私は、特許済水鉱圏を含むら記訂正後の明細部を検討し、 内容を理解していることをここに表明します。	I hereby state that I have reviewed and understand, the contents of the above identified specification, including the claims, as amended by any amendment referred to above.	
党は、運动域則に具第37加第1条56項に定義されるとおり、帯音状体の対量について重要な情報を提示する最高があることを認めます。	I acknowledge the duty to disclose information which is material to patentability as defined in Title 37, Code of Federal Regulations, Section 1.56.	

omeint of time you are required to complete this form should be sent to the Chief Information Officer. Patent and Trademark Office, Washington SEND FEES OR COMPLETED FORMS TO THIS ADDRESS, SEND TO Commissioner of Patents and Trademarks, Washington, DC 20231.

# Japanese Language Declaration (日本語宣言書)

松江、宋国出兵第3.5福1.1.9条(a)-(d) 領文は3.6.5条 (b) 役に基き下記の。 米 国以外の国の少なくとも一世国を指 迎している行序格の条約 3.6.5 (a) 頃に基デく国際出版。又 は外国での音音当難もしくは発明者証の出類についての外国 優先権をここに出張するとともに、優先権を主張している。 本出類の前に出着されて特許または発明者記の外国出類を以 下に、枠内をマークすることで、示しています。

# Prior Foreign Application(s)

外国 11-	での元行出版 312466(Pat.	Appln.)	Japan
	(Number) (출북)	····	(Country) (居名)
	(Number) (番号)		(Country) (図名)

利は、第35編米国任典119条(e)項に基いて下記の米 製神肝出離規定に記載された権利をここに主張いたします。

f? (Application No.) (Filing Date) (出願益号) (出類日) 1:3

松は、下記の米面に共第35編120条に基いて下記の米 団神許出願に記載された権利。 又は米国を指定している特許 協力条約365条(c)に基ずく権利をここに主張します。ま 『本出贈の各請求通三の内容が米国出典第35増112条 義。項スは特許協力条約で規定された方法で先行する米国特 諸盛頭に開示されていない限り、その先行米国出頭書提出日 以際で本出題者の日本国内または特許協力条約国際提出日ま での発制中に入手された、運邦規則法典第37編1条56項 で募棄された特許資格の有無に関する重要な情報について開 示義務があることを認識しています。

(Application No.)	(Filing Date)	
(出版금루)	(出期日)	
(Application No.)	(Filing Date)	
(出賦달루)	(出版日)	

私は、私自身の知識に基ずいて本直言書中で利が行なう表 明が真実であり、かつ私の入手した情報と私の信じるところ に基ずく表明が全て真実であると信じていること、さらに故 意になされた虚偽の表明及びそれと同等の行為は米国法典第 18編第1001条に基プき、罰金当たは拘禁、もしくはそ の両方により処罰されること、そしてそのような改意による 虚偽の声明を行なえば、出頭した、又は既に許可された特許 の有効性が失われることを認識し、よってここに上記のごと く宣誓を致します。

I hereby claim foreign priority under Title 35. United States Code, Section 119 (a)-(d) or 365(b) of any foreign application(s) for patent or inventor's certificate, or 365(a) of any PCT international application which designated at least one country other than the United States, listed below and have also identified below, by checking the box, any foreign application for patent or inventor's certificate, or PCT international application having a filing date before that of the application on which priority is claimed.

2/November/1999	Priority Not Claimed 優元権主要なし	
(Day/Month/Year Fried) (出類年月日)	= =	
(Day/Month/Year Filed) (出版年月日)		

I hereby claim the benefit under Title 35. United States Code, Section 119(e) of any United States provisional application(s) listed below

(Application No.) (Filing Date) (出題母母) (出贈日)

I hereby claim the benefit under Title 35. United States Code. Section 120 of any United States application(s), or 365(c) of any PCT International application designating the United States, listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States or PCT international application in the manner provided by the first paragraph of Title 35, United States Code Section 112, I acknowledge the duty to disclose information which is material to patentability as defined in Title 37. Code of Federal Regulations, Section 1.56 which became available between the filing date of the prior application and the national or PCT international filing date of application.

(Status: Patented, Pending, Abandoned) (現況: 特許許可済、孫建中、故棄済) (Status, Parented, Pending, Acandoned) (現況: 特許許可濟、保属中、放棄済)

i hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon

#### Japanese Language Declaration (三本語言言書)

李化伏: 魁に下記の発明者として、本出類に関する一切の 子成者を失義并有疑制に対して進行する中華士国だは代理人 として、下記の者を指名いたします。(弁理士、または代理 人の氏名及び登及番号を明記のこと)

(第三以降の共同発明者についても同様に記載し、署名をす

直接電話運絡先: (名前及び電話番号)

POWER OF ATTORNEY: As a named inventor. I hereby appoint the following attorney(s) and/or agent(s) to prosecute this application and transact all business in the Patent and Trademark Office connected therewith (list name and registration number).

苦酒等什完

, Ö

ること)

And I hereby agoint as principal automys: David T, Nikaido, Reg., No. 22,65; Charles M, Marmetsin, Reg., No. 52,85; George B, Omen, P., Reg., No. 22,731; Robert B, Murray, Reg., No. 22,290; E, Marcie Emas, Reg., No. 22,131; Daugust H, Golfshush, Reg., No. 31,315; Momics Lim Kim, Reg., No. 81,015; Charles Lim Kim, Reg., No. 31,00; Katren K, Costanino, Reg., No. 35,107; Marmet A, Poulos, Rile, Reg., No. 37,500; Katren K, Costanino, Reg., No. 35,107; Marmet A, Poulos, Rile, Reg., No. 37,500; Katren K, Costanino, Reg., No. 35,107; Marmet A, Poulos, Rile, Reg., No. 37,500; Katren K, Costanino, No. 37,500; Katren K, Costanino, Reg., No. 35,103; Sharon N, Klesner, Reg., No. 36,335; and Murra Cogra, Reg., No. 42,75; Brailey D, Oddischa, Reg., No. 43,073; and N, Alexander Mobile, Rev. No. 42,75;

Please direct all communications to the following address: ARENT FOX KINTNER PLOTKIN & KAHN, PLLC 1050 Connecticut Avenue, N.W., Suite 600 Washington, D.C. 20036-5339 Tel: (202) 857-6000; Fax: (202) 857-6395

(Supply similar information and signature for third and subsequent

O 一当たは第一発明者名 Full name of sole or first inventor Kouii Shiraishi 急引者の著名 日行 Inventor's signature Date October 12, 2000 注所 Shinagawa-ku, Tokyo, Japan Carrenson Japanese 私書霜 Post Office Address c/o FUJIKASUI ENGINEERING CO., LTD., 4-3, Higashigotanda 1-chome, Shinagawa-ku, Tokyo, Japan 第二共同発明者名 Full name of second joint inventor, if any Takayoshi Harimoto 第二共同発明者の署名 日付 Second inventor's signature /2 October 12, 2000 住所 Residence Shinagawa-ku, Tokyo, Japan 国票 Citizensnio Japanese 私審靈 Post Office Address C/O FUJIKASUI ENGINEERING CO., LITD., 4-3, Higashigotanda 1-chome, Shinagawa-ku, Tokyo, Japan

ioint inventors.)

第三共同発明者名		Full name or third joint inventor, if any
		Toshiaki Matsuoka
第三共同発明者の署名	日付	Third inventor's signature 本文 发 爱 October 12, 2000
<b>全</b> 所		Residence
·		Shinagawa-ku, Tokyo, Japan
国籍		Citizensnip
		Japanese
私書箱		Past Office Address C/O FUJIKASUI ENGINEERING CO., LTD., 4-3,
		Higashigotanda 1-chome, Shinagawa-ku, Tokyo, Japan
第四共同発明者名		Full name of fourth joint inventor, if any
		Naoki Fujihata
第四共同発明者の署名	日付	Fourth inventor's signature Date
		藤畑直蔭 October 12, 2000
生所		Residence
		Shinagawa-ku, Tokyo, Japan
国海"		Citizenshira
V)		Japanese
私書箱		Post Office Address
VD		
100		c/o FUJIKASUI ENGINEERING CO., LTD., 4-3,
S. Carlotte		Higashigotanda 1-chome, Shinagawa-ku, Tokyo, Japan
第五类同竞明者名		Full name of fifth joint inventor, if any
s		Akihiko Hongyou
第五六同発明者の署名	日何	Fifth inventor's signature Date
住所点		
C		Residence
Tand .		Shinagawa-ku, Tokyo, Japan
		Gitzensnip
私書箱		Japanese
は現場		Post Office Address
		c/o fujikasui engineering co., Ltd., 4-3,
		Higashigotanda 1-chome, Shinagawa-ku, Tokyo, Japan
第六共同発明者名		ful name of sixth joint inventor, if any Katsuo Oilkawa
第六共同発明者の署名	日付	Sixth inventor's signature
<b>住所</b>		Residence Shinagawa-ku, Tokyo, Japan
国等		Citizenarup
		Japanese
私事箱		Post Office Address C/O FUJIKASUI ENGINEERING CO., LTD., 4-3,
		Higashigotanda 1-chome, Shinagawa-ku, Tokyo, Japan

Tokyo, Japan Page 4 of 5

第七共同発明者名		Full name of seventh joint inventor, if say  Kazuo Takeda
第七共同発明者の署名	日付	Seventh inventor's signature 竹用一男 Date October 12, 2000
住所		Residence Shinagawa-ku, Tokyo, Japan
3# •		Citatenship Japanese
私書箱		Post Olific: Address c/o FUJIKASUI ENGINEERING CO., LITD., 4-3,
		Higashigotanda 1-chome, Shinagawa-ku, Tokyo, Japan
第八共同発明者名		Full name of eighth joint inventor, if any
第八共同発明者の署名	日付	Eighth inventor's signature Data
住所		Residence
		Cicitenship
私書館		Pari Office Addrass
- A		
第九共同発明者名		Full name of much joins leventor, if say
第九共同発明者の署名	日付	Ninth inventor's signaturo Osta
住所以		Residence
O# C		Civi exercibilge
私書箱		Past Offica Address
第十共同発明者名		full name of senth joint inventor, it any
第十共同発明者の署名	8付	Tenth inventor's signature Date
住所		Residence
<b>3</b> 76		Cittenship
私書箱		Pass Office Address